STATE UNIVERSITY OF MARINGÁ AGRICULTURAL SCIENCES CENTER

EFFECT OF CHEMICAL ADDITIVES ON THE CONSERVATION OF TROPICAL GRASS SILAGE

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MARINGÁ State of Paraná October – 2020

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Thesis presented to the Graduate Program in Animal Science of the State University of Maringá in partial fulfillment of requirements for MASTER OF SCIENCE degree IN ANIMAL SCIENCE, mainly: Animal Production.

> MARINGÁ State of Paraná October – 2020

Dados Internacionais de Catalogação-na-Publicação (CIP) (Biblioteca Central - UEM, Maringá - PR, Brasil)

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G633e	Gomes, Ana Luiza Mendonça Effect of chemical additives on the conservation of tropical grass silage / Ana Luiza Mendonça Gomes Maringá, PR, 2021. 52 f.figs., tabs.
	Orientador: Prof. Dr. João Luiz Pratti Daniel. Coorientador: Prof. Dr. Cloves Cabreira Jobim. Dissertação (Mestrado) - Universidade Estadual de Maringá, Centro de Ciências Agrárias, Departamento de Zootecnia, Programa de Pós-Graduação em Zootecnia, 2021. 1. Digestibilidade. 2. Nitrito de sódio. 3. Capim tropical - Silagem. 4. Aditivos químicos. 5. Nutrição animal. I. Daniel, João Luiz Pratti, orient. II. Jobim, Cloves Cabreira, coorient. III. Universidade Estadual de Maringá. Centro de Ciências Agrárias. Departamento de Zootecnia. Programa de Pós-Graduação em Zootecnia. IV. Título.
	CDD 23.ed. 636.085

Rosana de Souza Costa de Oliveira - 9/1366



UNIVERSIDADE ESTADUAL DE MARINGÁ CENTRO DE CIÊNCIAS AGRÁRIAS

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TITULAÇÃO: Mestre em Zootecnia - Área de Concentração Produção Animal

APROVADA em 29 de outubro de 2020.

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To my parents and sisters. My greatest examples and my foundation. Love unconditional and eternal gratitude.

AKNOWLEDGEMENTS

To God for always guiding my steps and allowing me to accomplish and unimaginable achievements.

To my parents Sueli de Fátima Mendonça Gomes and Gercio Grosso Gomes for support and guide me throughout all my life. Thank for all their love and encouragement. You are everything for me!

To my sisters, Jaqueline and Maria Clara, for their immense companionship and for being light in my life.

To Egídia, my grandmother and my treasure. Thank you for always believing in me and for all the prayers.

To my supervisor Prof. Dr. João Luiz Pratti Daniel, for sharing with me knowledge and experiences. Thank you for your patience and friendship, for contributing with my professional and personal development.

To Prof. Dr. Clóves Cabreira Jobim and the Silage Study Group and Hay, which helped me so much in this work.

To Iguatemi's Experimental Farm employees and the technicians of LANA laboratory for all help.

To the CNPq foundation (Conselho Nacional de Desenvolvimento Científico e Tecnológico) for the master's scholarship,

BIOGRAFY

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ABSTRACT

As producing tropical grass silage with high feeding value is still a challenge, we examined the effectiveness of sodium nitrite-based additives on guinea grass (Megathyrsus maximum cv. Mombaça) silage quality. The forage was mechanically harvested from four 3ha fields and divided into 5 piles per field to receive one of the following treatments (fresh matter basis): no additive (control), soybean hulls (100 g/kg; SH), sodium nitrite (1 g/kg; NIT), sodium nitrite (1 g/kg + hexamine (0.65 g/kg; NIT+HEX), and formic acid (85%) (4 mL/kg; FA). The SH and FA were included as positive controls. Sodium nitrite-based additives (sodium nitrite applied alone or in combination with hexamine) and FA were capable of decreasing clostridial development, resulting in lower concentrations of NH₃-N and n-butyric acid and reduced dry matter (DM) loss during fermentation, whereas protein quality and hygienic quality (reflected by lower *Clostridium* counts) were improved. A strong linear relationship was detected between the concentrations of butyric and valeric acids and DM losses during fermentation ($R^2 = 0.87$, P < 0.01). Addition of SH improved DM digestibility and slightly decreased fermentation losses, but it did not lead to butyricfree silages. The use of sodium nitrite-based additives was effective to improve the fermentation quality of tropical grass silage, and its combination with hexamine was superior to the sole use of sodium nitrite. All treatments improved in vitro dry mater digestibility over untreated silage. Only NIT, NIT+HEX and FA increased the concentration of rumen-undegradable protein, with NIT+HEX and FA outperforming NIT.

I. INTRODUCTION

1. Literature Review

1.1. Tropical grass silage

Tropical grasses are perennial or semi-perennial crops with high dry matter (DM) yield (e.g., 20–30 t/ha) and great potential as silage source (Pereira, 2004; Daniel et al., 2019). However, harvesting tropical grass with high nutritive value may result in excess of moisture and low soluble carbohydrates content and can impair the ensiling process (Bernardes et al., 2018; Silva et al., 2019). The mal fermentation may result in silages with high fermentative losses and low hygienic quality (McDonald et al, 1991).

Wilting is an interesting strategy to increase DM and improve silage fermentability in short and thin-stemmed crops, but wilting tall and thick-stemmed grasses is a challenge. Addition of absorbent ingredients (e.g. soybean hulls) also are used as way to decrease moisture content and supply nutrients to promote the development of beneficial -microorganisms and to enhance fermentation (McDonald et al., 1991).

1.2. Fermentability coefficient

Weissbach et al. (1974) coined the concept of fermentability coefficient (FC), that considers the contents of DM, soluble carbohydrates (SC) and buffering capacity (BC) of the pre-ensiled material [FC = DM (g/kg FM) + 80 × SC (g/kg DM) / BC (g/kg DM)]. Based on this model, forages with FC above 450 might result in well-fermented silages, while forages with FC below 350 are prone to fermentation problems. Nonetheless, in a further development of models to predict the run of fermentation, Weissbach and Honig (1996) reported that 51% of silages with FC > 350 underwent butyric fermentation, which indicate that the FC as a sole index is not capable of predicting the run of fermentation. Yet, according the authors, LAB counts and nitrate concentration in the fresh crop also have a great effect on crop ensilage process. They inferred that materials with LAB counts below 10^5 LAB/g fresh matter (FM) and nitrate content below 0.5 g NO₃/kg DM have a high risk of clostridium development, regardless the FC. Under extensive farming management (low fertilizer input), Weissbach et al. (1993) also reported that 77% of grass silages produced with FC > 350 underwent butyric fermentation, regardless of LAB count. Those findings indicate that nitrate concentration at ensiling has a pronounced effect on the run of fermentation, as will be discussed below.

1.3. Nitrate and grass silage fermentation

Nitrate occurs naturally in green forages and its concentration depends of the vegetal organs of plants, growth conditions and N fertilization level (Weiss et al., 2006). Nitrate has important significance on the silage fermentation process (Wieringa, 1966) and it is necessary to be considered to estimate the ensiling potential of grasses. Nitrate acts as a natural inhibitor of clostridia, mainly during early phases of the fermentation process (Spoelstra, 1985). Optimum nitrate levels range from 4.4 to 13 g NO₃/kg DM (Wieringa, 1966; Kaiser and Weiss, 2004). Lower levels may not be enough to inhibit undesirable microorganisms (e.g. clostridia), whereas excessively high levels might buffering the silage and impair pH drop. When nitrate is present in grass within such optimal range, the prediction of minimal DM content [DMmin (g/kg FM) = $450 - 80 \times$ SC/BC] developed by Weissbach et al. (1974) to prevent butyric fermentation is still valid. However, in grasses harvested with lower nitrate levels, the FC estimated by Weissbach et al (1974) is not suitable for rating the ensiling potential.

Considering the relatively high proportion of grasses with an enough FC (>350) that occasionally underwent butyric fermentation, Kaiser et al. (2002) developed new models to estimate the ensiling potential of temperate forages with low nitrate content. In their models, the minimum DM to obtain butyric-free silages was define as follows:

crops with low clostridial contamination at ensiling (<10³ cfu/g FM), DM_{min} (g/kg FM) = $620 - 71 \times SC/BC$; and crops with high clostridial contamination at ensiling ($\geq 10^3$ cfu/g FM), DM_{min} (g/kg FM) = $890 - 130 \times SC/BC$. From those models, the DM required to prevent mal fermentation would exceed the acceptable values (e.g. 400 to 600 g/kg FM) to store forage as silage without increase the risk of aerobic deterioration, especially in horizontal-unwalled silos (drive over silos). Therefore, the adoption of chemical additives or the combination of wilting and chemical additives might be a reasonable strategy to prevent storage losses and produce grass silages with high feeding value.

1.4. Fermentation improvers

The main action of chemical additives is through the suppression or inhibition of undesirable microorganisms, such as clostridium, enterobacteria and listeria (Auerbach and Nadeau, 2019). The effect of formic acid, nitrate and hexamine will be discussed below.

1.4.1. Formic acid

During silage conservation, formic acid is used as dual purpose. It causes a drop in pH by direct acidification and has action on suppression of undesired spoilage bacteria (Auerbach and Nadeau, 2019). The immediate reduction in pH restricts activities of acid-sensitive epiphytic bacteria such as enterobacteria and aerobes, thus establishing conditions for LAB to develop quickly and dominate the silage (Kung et al., 2003). Additionally, the non-dissociated molecule of formic acid has a specific inhibitory effect, penetrating cell membrane by diffusion, releasing H + and acidifying the cytoplasm (Wooldford, 1975, Krebs et al., 1983, Lambert and Stratford, 1999, Warth, 1991).

Formic acid has been used as an additive in crops with low DM and sugar concentrations, in order to decrease pH rapidly (<4.2) and prevent clostridia development (Nadeau et al., 2000). As a result, formic acid induces water-soluble

carbohydrates and true protein preservation during ensiling and increase microbial protein synthesis in the rumen (Jaakkola et al., 2006, Muck et al., 2018). The intensity of declining pH post formic acid application is heavily dependent of dose level, water concentration and buffering capacity of forage crops (Kung et al., 2003). Application of a dose around 4 L/t FM of formic acid 85% has been common in countries with tradition in chemical additives (Auerbach and Nadeau, 2019).

Nowadays, the use of formic acid has decreased due to corrosive potential, which may cause occupational issues such as skin, eyes and lung irritations. In addition, it can occur damage to the machinery used in the silage process (Kung et al., 2003; Auerbach and Nadeau, 2019).

1.4.2. Sodium nitrite

Sodium nitrite (NaNO₂) is a reactive molecule with bacteriostatic and bactericidal activity. The main function of sodium nitrite is the suppression of undesirable spoilage bacteria, particularly at low pH levels (Woolford, 1975).

The antimicrobial action of nitrite is attributed to reactions associated with the generation of nitric oxide (NO) and probably peroxynitrite (ONOO–) and peroxynitrous acid (ONOOH), which are strong oxidizing species in vivo (Majou and Christieans, 2018). The latter two compounds may oxidase zinc fingers, protein thiols, membrane lipids, cysteine and arginine biosynthesis and iron-sulfur proteins, and DNA bases, producing DNA strand breaks, which results in NAD+ and ATP consumption. The phosphoroclastic system is a major pathway to ATP synthesis in many clostridia and depends on the activity of two iron-sulfur enzymes, ferredoxin and pyruvateferredoxin oxidoreductase. These enzymes act in the transport of electrons linked to the ATP production from pyruvate. Nitric oxide inhibits the iron-sulfur enzymes (which are associated to electrons transport and ATP production) due to the formation of iron-NO complexes such as catalase, ferrochelatase and aconitase. The inhibitory action in the clostridium development is for the commitment of respiratory activity, inhibiting them (Carpenter et al, 1987; Majou and Christieans, 2018). Sodium nitrite has a wide antimicrobial spectrum and showed a marked increase in activity with a reduction in

pH. On the other hand, yeasts demonstrated a high resistance to it, even at pH 4 and 3 (Wooldford, 1975).

1.4.3. Hexamine

Hexamethylene tetramine (hexamine) is a bacteriostatic agent and in acidic conditions, gradually decomposes in ammonia and formaldehyde (Restani et al., 1992). The formaldehyde reduces protein degradability by forming crosslinks between protein chains and have antimicrobial properties due to the ability to inactivate certain macromolecules, such as protein and nucleic acids (Woolford, 1975, Aurelli et al., 2011).

1.4.4. Combination of sodium nitrite and hexamine

The aim of combining nitrite and hexamine is to protect all fermentation phases. Nitrite has an action during early phases whereas the formaldehyde derived from hexamine has an antimicrobial action posteriorly, after pH drops. Nitrite plus hexamine improves the silage quality when compared with use of sole sodium nitrite, and some trials showed a synergetic effect of hexamine and sodium nitrite (Hellberg, 1967, Auerbach and Nadeau, 2019). The combination of nitrite and hexamine have the potential to improve the fermentation pattern, reducing butyric acid production and proteolysis during fermentation, resulting in silages with lower dry matter losses, desirable hygienic quality and feeding value (Weissbach and Auerbach, 2012).

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1 II. Effect of sodium nitrite-based additives on the conservation and nutritive

2

value of guinea grass silage

3 (Manuscript formatted according to Animal Feed Science and Technology)4

5 Abstract

As producing tropical grass silage with high feeding value is still a challenge, we 6 7 examined the effectiveness of sodium nitrite-based additives on guinea grass (Megathyrsus maximum cv. Mombaça) silage quality. The forage was mechanically 8 9 harvested from four 3-ha fields and divided into 5 piles per field to receive one of the 10 following treatments (fresh matter basis): no additive (control), soybean hulls (100 g/kg; SH), sodium nitrite (1 g/kg; NIT), sodium nitrite (1 g/kg + hexamine (0.65 g/kg; 11 NIT+HEX), and formic acid (85%) (4 mL/kg; FA). The SH and FA were included as 12 13 positive controls. Sodium nitrite-based additives (sodium nitrite applied alone or in combination with hexamine) and FA were capable to increase curtailing clostridial 14 15 development, resulting in lower concentrations of NH₃-N and n-butyric acid and 16 reduced dry matter (DM) loss during fermentation, whereas protein quality and hygienic quality (reflected by lower *Clostridium* counts) were improved. A strong linear 17 relationship was detected between the concentrations of butyric and valeric acids and 18 DM losses during fermentation ($R^2 = 0.87$, P < 0.01). Addition of SH improved DM 19 digestibility and slightly decreased fermentation losses, but it did not lead to butyric-20 21 free silages. The use of sodium nitrite-based additives was effective to improve the 22 fermentation quality of tropical grass silage, and its combination with hexamine was superior to the sole use of sodium nitrite. All treatments improved in vitro dry mater 23 digestibility over untreated silage. Only NIT, NIT+HEX and FA increased the 24 concentration of rumen-undegradable protein, with NIT+HEX and FA outperforming 25 NIT. 26

Keywords: fermentation quality, formic acid, hexamine, sodium nitrite, soybean hulls,tropical grass

29 *Abbreviations*: ADF, acid detergent fiber expressed inclusive of residual ash; aNDF,

30 neutral detergent fiber assayed with a heat stable amylase and expressed inclusive of

- residual ash; BC, buffering capacity; CFU, colony-forming units; CON, control; CP,
- 32 crude protein; DM, dry matter; DM_{min}, minimum dry matter; FA, formic acid; FC,
- 33 fermentability coefficient; NH₃-N, ammonia nitrogen; NH₃-N_{corr}, ammonia nitrogen
- 34 corrected for addition of nitrogen by additives; LAB, lactic acid bacteria; NIT: sodium

nitrite; NIT+HEX: sodium nitrite plus hexamine; RDP: rumen degradable protein; RUP:

rumen undegradable protein; SC: soluble carbohydrates; SD, standard deviation; SE,

37 standard error of the mean; SH, soybean hulls.

38

1 Introduction

Tropical grasses are perennial or semi-perennial crops with high dry matter (DM) yield (e.g., 20-30 t/ha per year) high regrowth vigor and high adaptation to different climate and soil conditions resulting in lower agronomical risks (Da Silva et al., 2009; Jank et al., 2010). Therefore, tropical grasses have huge potential as ensiled forage source. Although research on silage production from tropical grasses is not recent (Condé, 1970), achieving high quality tropical grass silage at farm level remains a challenge (Daniel et al., 2019).

In Brazil, approximately 24% of dairy farms (Bernardes and do Rêgo, 2014) and
9% of beef feedlots utilize tropical grass silage (Pinto and Millen, 2018), with *Megathyrsus* and *Urochloa* genera predominating. In the meantime, consultants have
claimed that tropical grass silage usage has declined in the last decade, in some way
because of the struggles faced during mechanical harvesting but mainly due to high
storage losses and low feeding value caused by undesirable fermentations, and too a late
harvest beyond the optimal stage of maturity.

Although tropical grass silages can be used as a second forage source in addition 16 17 to whole-plant corn silage in dairy diets (Daniel et al., 2019), only very recently has its 18 use attracted attention by beef farmers, mainly due to the backgrounding intensification 19 (from weaning until the start of the finishing period). In growing rations, the use of tropical grass silage enables nutritionists to formulate diets that balances energy intake, 20 21 thereby avoiding that cattle gain weight too quickly, which negatively impact carcass weight as animals reach their mature size (Owens et al., 1995; Taylor et al., 2015). Also, 22 23 dairy heifers might be raised under controlled energy diets, which will reduce fat deposition and avoid negative effects on the future lactation performance (Drackley, 24

25 2008). Therefore, this demand now requires resumption of research to find effective26 strategies to solve the problem of poor fermentation quality in tropical grass silage.

27 Current strategies used to improve the fermentation in tropical grass silages have not led to consistent results. Although feasible, wilting is challenging in tall and thick-28 stemmed tropical grasses during the summer. Addition of absorbents consistently 29 decreases effluent formation but turns the ensiling process more complex in large 30 operations. Moreover, improving crop ensilage process through a sole increase in DM 31 content has not always guaranteed good silage preservation (Weissbach et al., 1993; 32 Weissbach and Honig, 1996; Kaiser et al., 2002). Inoculation with lactic acid bacteria 33 34 (LAB) have improved fermentation in some studies (Santos et al., 2014), but their 35 benefits are inconsistent mainly when the crop had a high moisture content at ensiling (Tomaz et al., 2018; Gouvea et al., 2020). 36

37 Chemical additives are widely used for silage making from temperate grasses. For a long time since the mid 20th century, the use of formic acid dominated in countries with 38 long tradition in silage making, e.g. Scandinavian countries, but its corrosivity on metal, 39 40 skin and eyes damages machinery and poses serious health risks to workers, leading to a 41 decline in popularity (Auerbach and Nadeau, 2019). Alternatively, chemical additives 42 based on sodium nitrite were introduced to the market place in the mid 1980ies, which 43 have the ability to inhibit butyric fermentation without the concerns reported for formic acid (Weissbach et al., 1989a; Weissbach and Auerbach, 2012). However, to the best of 44 45 our knowledge there is no information available about the effects of sodium nitritebased additives in tropical grass silage. Thus, the objective of this study was to examine 46 47 the effectiveness of sodium nitrite-based additives on fermentation, aerobic stability and nutritive value of guinea grass silage. We hypothesized that, as known for temperate 48 grasses, sodium nitrite-based additives would elucidate the same effects as formic acid 49

50	also in tropical grasses and could therefore be a useful and a practical management tool
51	to ensure high silage quality.

52 Material and Methods

53

54 *Ensiling and sampling*

Four 3-ha fields of guinea grass (Megathyrsus maximus cv. Mombaca), which 55 had been in use for three years at the Estância Independente Farm (Mandaguari, PR, 56 57 Brazil), were harvested with a double-chop flail pull-type forage harvester (CFC-1800, 58 Casale, São Carlos, Brazil) on February of 2019, after 45 d of regrowth. At harvest, 59 canopy and stubble heights were 120 cm and 15 cm, respectively. The fields received annual nitrogen fertilization by urea at 200 kg/ha and by cattle slurry (30 to 40 m^3 /ha). 60 61 The chemical and microbiological composition of guinea grass at harvest is shown in 62 Table 1.

Direct cut chopped forage from each of the four field replicates was divided into 63 64 five piles (1.5 kg per pile). Subsequently, each pile received one of the following 65 treatments (fresh matter basis): no additive (control; CON), soybean hulls (100 g/kg; 66 SH), sodium nitrite (1 g/kg; NIT), sodium nitrite (1 g/kg) + hexamine (0.65 g/kg; NIT+HEX), and formic acid (85%, 4 mL/kg; FA). The SH and FA treatments were 67 68 included as positive controls, as soybean hulls have been widely used to increase DM content of tropical grass at ensiling, and FA has been considered a standard additive to 69 70 prevent *Clostridium* development, to preserve protein and to secure the hygienic quality 71 of temperate grass silages. All dilutions were made using the same volume of water (15 72 mL/kg) to avoid that the chemical additive use does not affect the DM content at 73 ensiling. Control and SH treatments also received the same volume of water. Samples

from each pile were collected for analysis of DM, soluble carbohydrates (SC), and
buffering capacity (BC).

76 Treated forage was then ensiled in vacuum sealed nylon-polyethylene bags ($33 \times$ 77 45 cm, 160 µm thickness), with 1 kg per bag and with 4 replicates per treatment, totaling to 20 silage bags. After 101 d of storage in a closed barn (16 to 32°C), silos 78 were weighed to calculate the fermentation losses. Gas loss was computed as the mass 79 loss in proportion of the DM ensiled, whereas DM loss was calculated as the difference 80 between the amount of DM ensiled and DM recovered in proportion of DM ensiled. 81 Silage samples (170 g) were collected for drying at 60°C for 72 h and to prepare 82 83 aqueous extracts (30 g of silage + 270 g of distilled water, blended for 2 min and 84 filtered through four layers of cheese cloth). The remaining material (800 g) was used to determine aerobic stability. 85

For the aerobic stability test, silages were transferred to 3-L plastic buckets with 86 a data logger placed in the center of the silage mass. Two additional data loggers were 87 set to record room temperature (25 ± 1.8 °C). The data loggers were programmed to 88 record the temperature of the room and the silages every 15 min for 10 d. Aerobic 89 90 stability based on temperature rise was defined as the time elapsed until silage 91 temperature reached 3°C above the room temperature (Honig, 1990). On each morning 92 of the 10 d aeration period, sub-samples (10 g) were collected from each bucked, approximately 12 cm from the top surface, for measuring silage pH during the aerobic 93 94 exposure. The pH was measured in aqueous extract prepared with 10 g of silage + 90 gof distilled water blended for 2 min and filtered through four layers of cheese cloth. 95 96 Aerobic stability based on pH rise was defined as the time elapsed until silage pH increased by 0.5 unit. 97

99	Microbial counts (yeasts and molds, lactic acid bacteria (LAB) and clostridia)
100	were evaluated in a serial dilution of the aqueous extract. Microorganisms were
101	enumerated in Petri dishes with selective media. Malt Extract Agar (M137, Himedia®,
102	632 Mumbai, India) acidified to pH 3.5 with lactic acid was used for enumeration of
103	yeasts and molds, and De Man Rogosa and Sharp (7543A, Acumedia®, Lansing,
104	Michigan, USA) was used for enumeration of LAB. The plates were incubated
105	aerobically at 30°C for 48 h before counting yeasts and LAB and for 72 h before
106	counting molds. Reinforced Clostridium Agar (M154, Himedia®, 632 Mumbai, India)
107	supplemented with neutral red and D-cycloserine (Jonsson, 1990) was used for
108	enumeration of clostridia after 5 d of incubation at 37°C. Colony-forming units (CFU)
109	were counted and expressed as log10.
110	Fresh crop samples, dried and ground at 1-mm, were used for measuring the
111	concentration of soluble carbohydrates (SC) by the phenol-sulfuric method (Hall, 1999),
112	buffering capacity (BC, g lactic acid/ kg DM; Weissbach, 1967), and nitrate (Beutler et
113	al., 1986). Fermentability coefficient (FC) and minimum DM content (DMmin) required
114	to obtain butyric acid-free silage were calculated as follows: $FC = DM (g/100 g FM) +$
115	$8 \times$ SC/BC (Weissbach et al., 1974); DMmin Weissbach (g/kg) = $450 - 80 \times$ SC/BC
116	(Weissbach et al., 1974); DMmin Kaiser (g/kg) = $620 - 71 \times SC/BC$ (Kaiser et al.,
117	2002; for crop with ≤ 1 g NO3/kg DM and low count of clostridium spores).
118	Silage sub-samples were dried and ground (1-mm, Wiley mill) and analyzed for DM
119	(AOAC, 1990), ash (AOAC, 1990), crude protein (CP; AOAC, 1990), neutral detergent
120	fiber (aNDF; assayed with a heat stable amylase and sodium sulfite and expressed
121	inclusive of residual ash; Mertens, 2002), acid detergent fiber (ADF; assayed
122	sequentially and expressed inclusive of residual ash; Van Soest, 1973), soluble

123	carbohydrates (Hall, 1999), soluble protein (Krishnamoorthy et al. 1982) and acid-
124	detergent insoluble nitrogen (ADIN) and neutral-detergent insoluble nitrogen (NDIN)
125	contents (Licitra et al., 1996). Nitrogen fractionation was determined according to
126	CNCPS v. 6.5 (A1, A2, B1, B2 and C fractions) (Van Amburgh et al., 2015). From
127	nitrogen fractionation, the proportion of rumen degraded protein (RDP) and rumen
128	undegraded protein (RUP) were calculated using the first-order approach $[kd/(kd + kp)]$
129	for a mid-lactation dairy cow (Van Amburgh et al., 2015). Fractional passage rates
130	(liquid, concentrate and forage) were estimated assuming 20 kg/d DM intake, 50%
131	dietary forage level and 600 kg shrunk body weight (Tylutki et al., 2008).
132	A sample of undiluted aqueous extract was centrifuged (10,000 x g for15 min)
133	and the supernatant was used for analyses of fermentation end-products. The contents of
134	lactic acid (Pryce, 1969) and ammonia (Chaney and Marbach, 1962) were determined
135	by colorimetry. For treatments containing nitrite-based additives, ammonia
136	concentration was corrected considering that the conversion of nitrite and hexamine to
137	ammonia, were 50 and 90%, respectively. The other fermentation products were
138	analyzed by gas chromatography (GCMS 628 QP 2010 plus, Shimadzu, Kyoto, Japan)
139	using a capillary column Stabilwax, Restek, 629 Bellefonte, PA; 60 m, 0.25 mm ø, 0.25
140	μ m crossbond arbowax polyethylene glycol). The DM content was corrected for loss of
141	volatile compounds during oven drying (Weissbach and Strubelt, 2008).
142	

143 *Statistical analysis*

144 Statistical analyses were performed using the MIXED procedure of SAS 145 (version 9.4, SAS Institute, Cary, NC, USA). The model included a random effect of 146 field and a fixed effect of treatment. Treatment means were compared by the Tukey-147 Kramer test ($\alpha = 0.05$ and $\alpha = 0.10$). Silage pH during aerobic exposure was analyzed as

148	repeated measures. The interaction between treatment and silo was used as error term.
149	The covariance structure based on the lowest corrected Akaike information criterion
150	was unstructured (UN). The relationships between silage characteristics were evaluated
151	using the REG procedure of SAS.
152	
153	Results
154	Forage characteristics
155	Fermentability traits of the forages after additive application are shown in Table
156	2. The inclusion of SH increased the DM content compared with the other treatments.
157	The SC content was similar among treatments, whereas BC was highest for NIT+HEX
158	followed by NIT. The fermentability coefficient was greatest for SH treatment. The
159	DM_{min} required to obtain butyric acid-free silage estimated by the equations of
160	Weissbach et al. (1974) and Kaiser et al. (2002) was highest for NIT and NIT+HEX but
161	they did not differ from CON and SH, whereas FA application resulted in the lowest
162	value of DM _{min} .
163	
164	Silage characteristics
165	Silage microbial counts, fermentation profile and aerobic stability are shown in
166	Table 3. The counts of LAB and yeasts were lowest in untreated guinea grass silage and
167	the addition of NIT, NIT+HEX and FA resulted in lower Clostridium counts when
168	compared with CON and SH. Molds were always below the detection limit of log 2
169	CFU/g, regardless of additive treatment.
170	The pH was higher in SH and NIT+HEX than in FA silage, and no differences
171	were detected between NIT+HEX and NIT and CON. The FA application led to the
172	lowest pH, which did not differ from CON. All fermentation products were affected by

additive treatment, except ethyl acetate, 1, 2-propanediol, ethyl lactate and propyl acetate. 173 174 When compared with CON, total NH₃-N was decreased by SH and chemical additives, 175 with the lowest value observed in FA silage. Correction for nitrogen applied with the sodium nitrite-containing additive resulted in the treatment NIT+HEX with the lowest 176 177 concentration of NH₃-N_{corr}. Silage treated with FA and NIT+HEX had the highest concentrations of lactic acid, whereas NIT had an intermediate value and did not differ 178 179 from SH. The control had the lowest concentration of lactic acid, but it was similar to 180 SH. The use of FA decreased the acetic acid concentration when compared with CON, SH and NIT. 181

182 The n-butyric acid concentration was greater in CON followed by SH. Silages 183 treated with NIT+HEX, FA and NIT had low concentrations of this compound (<3 g/kg 184 DM). The 2,3-butanediol and propionic acid were higher in CON followed by SH, 185 whereas the chemical additives depressed the formation of those compounds. Greater 186 concentrations of i-butyric, i-valeric and n-valeric acids were found in SH followed by CON. The NIT+HEX and FA were the only treatments capable of suppressing the 187 formation of n-propanol. Untreated silage had the highest undissociated VFA 188 189 concentration, followed by NIT and SH, whereas FA and NIT+HEX resulted in the 190 lowest undissociated VFA concentration. 191 When compared with CON, all additives reduced gas loss, with lower values for

191 when compared with CON, an additives reduced gas loss, with lower values for 192 chemical additives than SH, and NIT+HEX being more effective than FA and NIT. 193 Untreated silage showed the highest DM loss during fermentation and the use of 194 NIT+HEX caused the largest reduction of all additives. A positive linear relationship 195 ($R^2 = 0.87$, P < 0.01) was detected between the sum of i-butyric, n-butyric, i-valeric and 196 n-valeric acids and the DM loss during fermentation (Figure 2). Including the traits n-197 propanol, 2,3-butanediol and propionic acid in the regression model in addition to

butyric and valeric acids improved the relationship ($R^2 = 0.88$, P < 0.01, RMSE = 10.0, 198 199 $Y = 47.2 + 3.08 \times X$). No temperature increase was observed in any of the silages 200 during the entire duration of the aerobic stability test of 10 d. However, in silages 201 treated with NIT+HEX, NIT and FA pH increased after 6 or 7 d of aerobic exposure (Figure 1). A positive linear relationship ($R^2 = 0.55$, P < 0.01) was detected between the 202 203 ratio of total undissociated VFA concentration and the sum of SC and lactic acid 204 content and the aerobic stability based on pH rise (Figure 3). 205 Data about chemical composition and nutritive value of guinea grass silage is 206 presented in Table 4. Treatment SH showed the highest DM and the lowest ash 207 concentrations. Treatment with FA decreased aNDF content, but this did not differ from 208 SH, and a trend was observed for lower NDF concentration in FA silage when 209 compared with CON (P = 0.07). Regardless of additive type, all chemically treated 210 silages had a lower ADF content when compared with CON and SH. The SC 211 concentration was higher in silages treated with chemical additives, with the highest value observed in FA followed by NIT+HEN and then NIT. Untreated silage had the 212 lowest CP content, whereas SH addition increased it over all treatments. Chemical 213 214 additives decreased the proportion of soluble CP, with the lowest values detected in 215 NIT+HEX and FA. The N fractionation revealed that CON had a greater proportion of 216 A1 and the lowest proportions of B2 and C fractions among all treatments. Addition of SH increased the proportion of A2 (non-ammonia soluble protein) over silages treated 217 218 with sodium nitrite-containing additives. In addition, it also reduced the proportions of B1 when compared with NIT and of B2 (insoluble potentially digestible protein) over 219 220 NIT, NIT+HEX and FA. The highest proportion of fraction C (indigestible protein) was detected in NIT+HEX silages. The proportion of RDP was greatest in CON, followed 221

by SH and NIT, whereas the use of NIT+HEX and FA resulted in the largest proportionof RUP.

The IVDMD was increased by SH addition and all chemical additives. Recovery of digestible DM was improved over CON by SH and NIT, and the highest value was detected in silage treated with NIT+HEX and FA.

227

228 Discussion

229

230 Crop characteristics, fermentation pattern, dry matter loss and aerobic stability of

231 guinea grass silage

232 Based on the evaluated chemical and microbiological traits of guinea grass at 233 ensiling, the forage can be considered of typical composition. Our DM data at ensiling 234 and nutrient composition substantiate previous observations by Kotha et al. (2018), and 235 by Tomaz et al. (2018), who monitored the chemical changes in guinea grass as affected by sward heights. The FC, which is used mainly to predict the risk of poor fermentation 236 quality caused by clostridia (Weissbach et al., 1974), in the untreated forage and in the 237 238 forage that received chemical additives were similar to the values reported by Tomaz et 239 al. (2018) for the sward height used in our study. The increased FC by SH inclusion can be explained by the higher DM caused by this treatment. Although the FC increase 240 241 induced by SH (from 34 to 39.5) slightly reduced DM losses, such improvement in FC 242 was not enough to prevent clostridial fermentation.

Beyond the FC, forage nitrate concentration plays a major role in inhibiting clostridia during the early fermentation stages due to its microbial degradation into nitrite and NO_x (nitric and nitrous oxides), which are potent clostridial inhibitors (Wieringa, 1966; Spoelstra, 1985; McDonald et al., 1991). Under extensive farming

247 management, with low input of fertilizers and, in turn, low nitrate content in crop,

248 Weissbach et al. (1993) reported that 77% of grass silages with FC > 35 underwent

249 butyric fermentation, regardless of LAB count.

More evidence, however, suggests an interaction between nitrate and epiphytic 250 251 LAB number based on silages produced from 244 different forages (including 195 from 252 grasses produced at different N fertilization levels) grown in a temperate climate 253 (Weissbach and Honig, 1996). These authors showed that the highest risk of butyric 254 acid formation (78% frequency of butyric acid-containing silages) existed when forage had $<10^5$ cfu epiphytic LAB/g and <0.5 g NO₃/kg at ensiling. When each factor was 255 256 considered individually, 26% of the silages made from forage high in nitrate (> 1 g/kg 257 DM) and 4% of the silages produced from forage with high epiphytic LAB count (> 10^6 258 cfu/g) were free of butyric acid. As the forage used in our study had an epiphytic LAB 259 count in excess of 10^6 cfu/g, the risk of poor fermentation quality should have been low 260 given that the observations from temperate grasses are also applicable to tropical species. However, this was not the case, as untreated silages clearly showed 261 characteristics typical for clostridial fermentations. Namihira et al. (2010) showed that 262 263 nitrate levels in guinea grass depended on N fertilization rate and that forage nitrate 264 concentration was strongly negatively related with n-butyric acid formation in the silage (r = -0.97).265

Despite the high epiphytic LAB count, which was much higher than found in guinea grass and Napier grass by Khota et al. (2018), CON and SH silages were prone to *Clostridium* development, as reflected by the presence of n-butyric acid at significant concentrations. Even higher butyric acid levels exceeding 30 - 50 g/kg DM in guinea grass silages have been reported previously but forage epiphytic LAB counts were not presented (Namihira et al., 2010; Tomaz et al., 2018). Likely, epiphytic LAB population

may not have a sufficiently high proportion of species that can convert plant sugar
efficiently into lactic acid (homofermentative LAB), thereby reducing the pH rapidly
and significantly preventing clostridial development. Additionally, LAB present on the
crop may not have been capable of thriving competitively due to excessively high
moisture content (Pahlow and Weissbach, 1999). Nevertheless, our findings are in line
with the lack of consistency with the application of inoculants containing LAB in
tropical grass silages (Igarasi, 2002; Ribeiro et al., 2009; Tomaz et al., 2018).

279 Models developed by Kaiser et al. (2002) for predicting the ensiling potential of temperate grasses indicate that for crops with low nitrate content (≤ 1 g/kg DM), the 280 281 minimum DM content required to obtain butyric acid-free silages is much higher than 282 that with adequate nitrate content. In our study, the guinea grass with a low nitrate content (0.173 g/kg DM) would have needed a $DM_{min} \ge 530$ g/kg to prevent 283 284 Clostridium development (Kaiser et al., 2002), which is challenging to attain in most 285 practical conditions with tall and thick stemmed tropical grass silages. Moreover, such 286 high DM level might be not recommended for practical silage making, due to the 287 increased risk of loss by aerobic deterioration (Wilkinson and Davies, 2013), mainly 288 when silage is intended to be stored in horizontal silos, especially in unwalled 289 horizontal silos.

In addition to high concentrations of butyric and valeric acids, untreated silages showed traits that are characteristic for *Clostridium* growth in silage made from tropical and temperate grasses as well as from alfalfa, e. g. low concentrations of lactic acid and relatively high acetic acid content as well as an elevated proportion of NH₃-N (Namihira et al. 2010; Tomaz et al., 2018; Auerbach et al., 2016). Although we only measured fermentation characteristics after an extended storage period, it is likely that, due to the very low nitrate concentration in the forage, *Clostridium* proliferation was supported

already from the early stages of fermentation, as was shown by Namihira et al. (2010).
Although Tomaz et al. (2018) showed good effects of using an absorbent (i.e., citrus
pulp) to improve guinea grass fermentation, the SH addition did not result in
satisfactory silage quality in our study. This finding, in addition to challenges to the
ensiling management, leads us to question the feasibility of this strategy to enhance
silage fermentation quality, especially for larger farm operations.

303 To our best knowledge for the first time on tropical grass, we confirmed observations from temperate forages that the application of sodium nitrite-containing 304 305 additives and formic acid are successful to suppress clostridial development for an 306 extended storage period with no difference between NIT+HEX and FA. Weissbach et 307 al. (1989b) and Reuter and Weissbach (1991) showed in a total of 143 trials performed between 1984 and 1989 that the combination of sodium nitrite (900 g/t) and hexamine 308 309 (600 g/t) was as effective as 4 L/t of formic acid (85%) in reducing the Clostridium 310 spore load and in restricting the metabolic activity of clostridia during fermentation. Obviously, in our study the combination of NIT+HEX was superior to NIT in terms of 311 fermentation process efficiency as reflected by DM losses and the extent of proteolysis 312 313 despite no, or not biologically relevant, differences in the concentrations of butyric and 314 valeric acids. Moreover, based on 14 trials with a total of 42 observations, Weissbach 315 (personal communication) showed that the individual use of sodium nitrite (1 g/kg) or 316 hexamine (0.6 g/kg) produced silage with higher mean butyric acid content and a lower 317 frequency of butyric acid free silages than was found for the combination of both chemicals used at the given application rate. This contradicts observations from one trial 318 319 by König et al. (2019) who found no additional benefit of combining sodium nitrite with graded doses of hexamine, but their forage had a high nitrate content (3.8 g/kg DM), 320

resulting in the formation of only small butyric acid concentrations (< 3 g/kg DM) even
when the forage did not receive any additive treatment.

323 Regardless of treatment, yeast numbers were higher at silo opening (> $\log 4.3$ cfu/g) than at ensiling (log 3.1 cfu/g) indicating a development during storage. This 324 325 observation was unexpected due to the presence of a high concentration of total VFA acids with antifungal activity (C_2 to C_5) in all silages although the composition of total 326 327 VFA differed between treatment, with CON and SH only containing significant quantities of butyric and valeric acids. According to Woolford (1975), the inhibitory 328 329 effect on yeasts increases with increasing chain length. We can only speculate why 330 yeasts could develop and survive during storage because there is no information 331 available when those acids were produced, either already during the initial stages of fermentation by conversion of plant sugar, or during the later phase by utilizing lactic 332 333 acid. Despite the relatively high yeast counts compared with the threshold value of 10^5 334 cfu/g signifying a high risk of aerobic instability (Jonsson and Pahlow, 1984), all silages 335 remained stable throughout the 10 d of air exposure when aerobic stability was evaluated using the temperature method by Honig (1990). As stated by Jonsson and 336 337 Pahlow (1990), strictly anaerobically stored silages may not heat-up even in the 338 presence of high yeast numbers when lactate-utilizing species were absent, which we 339 did not test. In support of this, da Silva et al. (2020) found a positive effect of a 340 chemical silage additive with antimycotic action on aerobic stability despite similar 341 yeast counts than that of untreated whole-plant corn silage, which received air treatment for 2 h/wk over 63 d of storage. They explained this finding with the additive effect on 342 343 the yeast population composition, causing a shift from non-lactate utilizing Candida pumulis in untreated silage to lactate-assimilating Pichia kudriavzevii in treated silage. 344 345 Additionally, more heat is required in wet silage than in drier silage (Wilkinson and

Davies, 2013), which, in turn, requires high concentrations of utilizable substrate (mainly sugar and lactic acid) to produce enough heat to detect a temperature increase. Recently, Auerbach and Nadeau (2020) found a strong positive relationship ($R^2 = 0.67$, P < 0.01) between the total concentration of water-soluble carbohydrates and lactic acid at silo opening and the extent of aerobic deterioration (based on cumulated temperature).

352 Although temperature remained unchanged during the aeration in all silages, 353 changes in another indicator for aerobic microbial activity (i.e., pH) were observed in treatments NIT, NIT+HEX and FA when compared with CON, resulting in at least 2.25 354 355 d earlier onset of aerobic instability based on a pH increase of ≥ 0.5 units. Previous 356 studies also reported the occurrence of aerobic deterioration without detection of heat production in tropical grass silage (Bernardes et al., 2007). Despite being undetectable 357 358 at silo opening, molds may have developed during air exposure due to its higher resistance to antimycotic VFA compared with yeasts (Woolford, 1975), but it cannot be 359 ruled out that also aerobic bacteria have played a role. The generally slower growth of 360 361 molds compared with yeasts may explain why it took at least 6 to 7.75 d (treatments 362 NIT, NIT+HEX and FA) to reach a pH, which was at least 0.5 units higher than that 363 measured at silo opening. Unfortunately, we did not measure mold counts during 364 aerobic exposure. More studies on tropical grass silage are warranted to better 365 understand the process of aerobic instability and aerobic deterioration, respectively, and 366 the microorganism causing pH increases without heat detection. The pH difference used can help signifying aerobic spoilage in tropical grass silage and may be a more suitable, 367 368 more reliable and more robust indicator for aerobic microbial activity than is heat production. The power of the relationship between the ratio of the total concentration of 369 undissociated VFA and utilizable substrate (soluble carbohydrates and lactic acid) and 370

the time elapsed to attain a pH increase by 0.5 unit offered a plausible explanation for 371 372 the findings in our study. The higher the pH-dependent concentration of antimycotic 373 VFA and the lower the quantity of substrates utilizable by fungi, the faster a pH 374 increase is detected. At a given concentration of undissociated VFA, a higher substrate 375 concentration will result in a smaller ratio and, thus, in a faster onset of aerobic 376 deterioration, likely leading to a greater extent of spoilage. More studies are needed to 377 confirm whether this index [the ratio of the total concentration of undissociated VFA 378 and utilizable substrate (soluble carbohydrates and lactic acid)] is suitable to predict 379 aerobic stability in other silage types (e.g., whole-plant corn silage, sorghum silage, 380 sugarcane silage, legume silages, high moisture corn, etc.).

381

382 Nutritive value of guinea grass silage

As the course of the fermentation process of tropical grass silage has been 383 384 unpredictable under research and farm conditions (Daniel et al., 2019), there is a high 385 risk of producing poorly fermented forages leading to poor silage intake and animal performance (Krizsan and Randby, 2007; Restle et al., 2003; Auerbach et al., 2012; 386 387 Santos et al., 2016). In this study, we demonstrated that the IVDMD of guinea grass 388 silage was improved by SH addition over untreated silage. However, it should not be ignored that digestible nutrients were added to the forage before ensiling by SH 389 390 supplementation. Considering the SH inclusion rate of 100 g/kg and an average of 80% 391 IVDMD in SH (Mohammadzadeh et al., 2007), the improvement of IVDMD over untreated silage (4.9 percentage units) can be explained simply by the SH addition. 392 393 Thus, the degradation extent of digestible nutrients process in the forage fraction of SH treatment was similar to that of untreated silage. On the contrary, all chemical additives 394 395 improved IVDMD and SC, which can be attributed to the protection of readily available

nutrients from degradation and the lower ADF concentration (Mills and Kung, 2012).

This assumption is supported by the much higher recovery of digestible DM after longstorage, especially in treatments NIT+HEX and FA.

Although CP concentration in tropical grasses is lower than in temperate species, it can 399 400 still be important and valuable to contribute to meeting the dietary protein requirements 401 of ruminant categories (e.g., growing cattle). Silva et al. (2009) showed that N fractions 402 in guinea grass depended on N fertilization rate and cutting heights, with protein soluble 403 fraction A decreasing with increasing N fertilizer application and cutting heights, 404 fraction B₁ remaining unaffected, and other fractions being altered at varying extent. To 405 our knowledge, this is the first study to show the effects of additives on N fractions in 406 guinea grass silage. Of all chemical additives, treatments NIT+HEX and FA reduced the 407 RDP concentration simultaneously increased the RUP fraction by about 61-62 g/kg CP, 408 which could have a significant effect on diet formulation and diet-related feeding costs. 409 In support of studies about positive effects of chemical additives on N fractions in silages from temperate grasses (Broderick et al., 2007; Nadeau et al., 2014; Nadeau et 410 411 al., 2015), these changes associated with the use of chemical additives may have the 412 potential to improve animal performance, but in vivo trials are warranted to test this 413 hypothesis and are considered of great scientific and commercial merit. Although 414 additional costs are incurred by silage additive use, a return-on-investment scenario could prove the use of chemical additives economically feasible. Considering the large 415 416 reduction of DM losses by NIT+HEX compared with untreated (71 g/kg DM), it seems that there is an opportunity by using additives such as NIT+HEX for ensiling tropical 417 418 grasses. More so, the effect of additives on protein quality (e.g., RUP) in guinea grass silage must be considered. Improved silage protein quality by additive use may enable 419 farmers to partially replace protein meals, and thus save feed costs. Overall, reducing 420

421 DM losses and improving protein quality may likely save more feed cost than the422 additive cost.

423

424 Conclusion

425 Formic acid and additives containing sodium nitrite alone or in combination with

426 hexamine were efficient in controlling clostridial fermentations in tropical grass silage,

427 reducing the fermentation losses, improving the nutritive value and securing the

428 hygienic quality of silages, whereas soybean hulls only slightly improved fermentation

429 and did not control *Clostridium* development. More studies on the effects of chemical

430 additives on tropical grass fermentation are warranted and animal studies are required to

431 evaluate their potential to reduce feed costs by partially replacing protein meals in

432 ruminant diets.

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atom additive employed on $(n - 1)$	Table 1 Chemical and microbiological composition of fresh guinea grass at harves						
efore additive application (n = 4) Item	Mean	SD					
DM ² , g/kg	247	4.8					
Crude protein, g/kg DM	77.1	1.16					
Soluble CP, g/kg CP	21.4	2.36					
aNDF ³ , g/kg DM	711	6.0					
ADF ⁴ , g/kg DM	385	5.1					
Ash, g/kg DM	109	1.7					
Soluble carbohydrates, g/kg DM	44.0	3.58					
Buffering capacity, g/kg DM	43.9	1.00					
Nitrate, g/kg DM	0.173	0.005					
Lactic acid bacteria, log cfu/g	6.11	0.032					
Clostridia, log cfu/g	2.48	0.231					
Yeasts, log cfu/g	3.10	0.128					
Molds, log cfu/g	3.12	0.239					

643 644 ²Dry matter. ³Neutral detergent fiber. ⁴Acid detergent fiber.

Treatment¹ Р-SEM² Item CON SH NIT NIT+HEX FA value DM^3 , g/kg 247^b 307^a 245^b 239^b 240^b 3.3 < 0.01 SC⁴, g/kg DM 42.3 43.9 42.2 43.2 48.2 1.96 0.23 BC^5 , g/kg DM 37.6^c 38.2^c 41.3^b 45.0^a 38.2^c 0.56 < 0.01 1.17^{ab} 1.11^{ab} 1.02^{b} 0.96^b SC:BC ratio 1.26^a 0.050 < 0.01 Fermentability 34.0^b 34.1^b 39.5^a 32.6^b 31.6^b 0.59 < 0.01coefficient⁶ 361^{ab} 356^{ab} 349^b DM_{min Weissbach}⁷, g/kg 368^a 373^a 4.0 < 0.01 537^{ab} 541^{ab} DM_{min Kaiser}⁸, g/kg 548^a 530^b 552^a 3.5 < 0.01

Table 2 Fermentability traits of guinea grass forage after additive application (n = 4)

646 ¹CON: without additive, SH: soybean hulls at 100 g/kg, NIT: sodium nitrite at 1 g/kg, NIT+HEX: Sodium

647 nitrite at 1 g/kg + Hexamine at 0.65 g/kg, FA: Formic acid 85% at 4 mL/kg.

648 ²Standard error of the mean.

649 ³Dry matter.

⁴Soluble carbohydrates.

651 ⁵Buffering capacity.

652 ${}^{6}FC = DM (g/100 g) + 8 \times SC/BC.$

⁷Minimum DM content to prevent butyric fermentation according to Weissbach et al. (1974; DM_{min}

654 $Weissbach = 450 - 80 \times SC/BC$).

⁸Minimum DM content to prevent butyric fermentation according to Kaiser et al. (2002; DM_{min Kaiser} =

656 620 - 71 × SC/BC; for crop with ≤ 1 g NO₃/kg DM and low count of clostridium spores).

657 a,b,c Tukey test ($\alpha = 0.05$).

		Treatment ¹					
Item	CON	SH	NIT	NIT+HEX	FA	SEM ²	<i>P</i> -value
Lactic acid bacteria, log cfu/g	6.71 ^b	7.55 ^a	8.17ª	8.12 ^a	8.20 ^a	0.172	< 0.01
Clostridia, log cfu/g	4.43 ^a	4.13 ^a	2.47 ^b	2.51 ^b	2.73 ^b	0.145	< 0.01
Yeasts, log cfu/g	4.27 ^b	5.15 ^a	5.20 ^a	5.19 ^a	5.46 ^a	0.146	< 0.01
Molds, log cfu/g	< 2	< 2	< 2	< 2	< 2	-	-
pH	4.60 ^{bc}	4.89 ^a	4.66 ^b	4.79 ^{ab}	4.44 ^c	0.053	< 0.01
NH ₃ -N, g/kg N	251 ^a	180 ^b	174 ^b	185 ^b	103 ^c	12.5	< 0.01
NH ₃ -N _{corr} ³ , g/kg N	251 ^a	180 ^b	138 ^{bc}	70.6 ^d	103 ^{cd}	12.4	< 0.01
Lactic acid, g/kg DM ⁴	2.03 ^c	9.15 ^{bc}	17.2 ^b	29.9 ^a	37.1 ^a	3.23	< 0.01
Acetic acid, g/kg DM	22.0 ^{ab}	15.9 ^{bc}	25.7 ^a	15.9 ^{bc}	10.8 ^c	1.56	< 0.01
n-Butyric acid, g/kg DM	16.9 ^a	11.2 ^b	2.33 ^c	1.95 ^c	2.23 ^c	1.164	< 0.01
Ethanol, g/kg DM	3.27 ^a	2.67 ^a	1.23 ^b	1.98^{ab}	1.15 ^b	0.327	< 0.01
2,3-Butanediol, mg/kg DM	2340 ^a	1531 ^b	527°	225°	826 ^c	164.1	< 0.01
Propionic acid, mg/kg DM	1838 ^a	1451 ^a	530 ^b	301 ^b	325 ^b	140.2	< 0.01
i-Butyric acid, mg/kg DM	571 ^b	1613 ^a	240^{bc}	56.7 ^c	253 ^{bc}	113.08	< 0.01
i-Valeric acid, mg/kg DM	135 ^b	243 ^a	61.8 ^{bc}	44.5 ^c	36.5 ^c	19.02	< 0.01
n-Valeric acid, mg/kg DM	102 ^b	208 ^a	49.8 ^{bc}	41.5 ^{bc}	34.0 ^c	16.15	< 0.01
n-Propanol, mg/kg DM	89.3 ^a	92.3 ^a	80.5 ^a	39.5 ^b	23.3 ^b	10.94	< 0.01
Ethyl acetate, mg/kg DM	14.3	16.5	13.5	19.0	13.0	3.87	0.80
1,2-Propanediol, mg/kg DM	2.50	2.67	2.25	2.25	2.50	0.781	0.99
Ethyl lactate, mg/kg DM	1.75	1.25	1.00	2.00	1.50	0.446	0.58
Propyl acetate, mg/kg DM	1.00	1.00	1.75	1.50	1.50	0.423	0.65
Undissociated VFA ⁵ , g/kg DM	25.3ª	13.6 ^{bc}	16.1 ^b	8.70 ^c	9.27 ^c	1.331	< 0.01
Gas loss, g/kg DM	90.9 ^a	69.6 ^b	57.6 ^c	46.0 ^d	54.4 ^{cd}	2.43	< 0.01
DM loss, g/kg DM	119 ^a	98.9 ^b	63.2 ^c	48.1 ^e	56.7 ^d	1.94	< 0.01
Aerobic stability T^6 , d	> 10	> 10	> 10	> 10	> 10	-	-
Aerobic stability pH ⁷ , d	$> 10^{a}$	9.25 ^{ab}	7.75 ^b	6.25 ^b	6.00^{b}	0.552	< 0.01

Table 3 Microbial counts, fermentation profile, aerobic stability and losses of guinea grass silages stored for 101 d (n = 4)

 1 CON: without additive, SH: soybean hulls at 100 g/kg, NIT: sodium nitrite at 1 g/kg, NIT+HEX: sodium nitrite at 1 g/kg + hexamine at 0.65 g/kg, FA: formic acid (85%) at 4 mL/kg.

- 661 ²Standard error of the mean.
- 662 ³NH₃-N corrected for addition of nitrogen by additives.
- 663 ⁴Dry matter.
- 664 ⁵ Sum of undissociated acetic, propionic, i-butyric, n-butyric, i-valeric and n- valeric acids.
- ⁶Aerobic stability based on temperature rise $(+2^{\circ}C)$. ⁷Aerobic stability based on pH rise (+0.5). 665
- 666
- ^{a,b,c,d}Tukey test ($\alpha = 0.05$). 667

668 **Table 4** Chemical composition and nitrogen fractions of guinea grass silages stored for

669

101 d (n = 4)

	Treatment ¹						
Item	CON	SH	NIT	NIT+HEX	FA	SEM ²	<i>P</i> -value
DM ³ , g/kg	211 ^c	282 ^a	226 ^{bc}	226 ^{bc}	234 ^b	4.15	< 0.01
Ash, g/kg DM	119 ^a	108 ^b	114 ^a	115 ^a	116 ^a	1.5	< 0.01
aNDF ⁴ , g/kg DM	717 ^x	708 ^{xy}	714 ^{xy}	709 ^{xy}	701 ^y	4.0	0.08
ADF ⁵ , g/kg DM	425 ^a	419 ^a	407 ^b	404 ^b	398 ^b	1.7	< 0.01
SC ⁶ , g/kg DM	8.92 ^d	8.85 ^d	11.3 ^c	13.8 ^b	19.0 ^a	0.62	< 0.01
CP ⁷ , g/kg DM	64.2 ^c	83.3 ^a	75.5 ^b	78.2 ^{ab}	77.6 ^{ab}	1.78	< 0.01
Soluble CP, g/kg CP	525 ^a	526 ^a	419 ^b	362 ^c	397 ^{bc}	9.4	< 0.01
IVDMD ⁸	0.562 ^b	0.611 ^a	0.596 ^a	0.608^{a}	0.615 ^a	0.006	< 0.01
RdDM ⁹ , g/kg DM	495 ^c	551 ^b	558 ^b	583 ^a	580 ^a	5.6	< 0.01
N fractionation ¹⁰ ,							
g/kg N							
A1	251 ^a	180 ^b	174 ^b	185 ^b	103 ^c	12.5	< 0.01
A2	274 ^{ab}	346 ^a	245b ^c	177 ^c	293 ^{ab}	18.0	< 0.01
B1	240 ^{ab}	230 ^b	277 ^a	243 ^{ab}	213 ^b	9.9	< 0.01
B2	151 ^c	154 ^c	212 ^b	287 ^a	297 ^a	10.4	< 0.01
С	83.4 ^b	89.9 ^{ab}	92.2 ^{ab}	109 ^a	93.2 ^{ab}	4.57	0.02
RDP ¹¹ , g/kg CP	749 ^a	733 ^b	716 ^c	687 ^d	686 ^d	3.7	< 0.01
RUP ¹² , g/kg CP	251 ^d	267 ^c	284 ^b	313 ^a	314 ^a	3.7	< 0.01

¹CON: without additive, SH: soybean hulls at 100 g/kg, NIT: sodium nitrite at 1 g/kg, NIT+HEX: sodium nitrite at 1 g/kg + hexamine at 0.65 g/kg, FA: formic acid (85%) at 4 mL/kg.

671 Intrite at 1 g/kg + nexamine at 0.05 g/kg, FA: formic acid (85%)
672 ²Standard error of the mean.

673 ³Dry matter.

674 ⁴Neutral detergent fiber.

675 ⁵Acid detergent fiber.

676 ⁶Soluble carbohydrates.

677 ⁷Crude protein.

678 ⁸*In vitro* dry matter digestibility.

⁹Recovery of digestible dry matter.

¹⁰Nitrogen fractionation according to CNCPS v. 6.5.

681 ¹¹Rumen degradable protein (calculated for growing cattle).

¹²Rumen undegradable protein (calculated for growing cattle).

683 a,b,c,dTukey test ($\alpha = 0.05$).

684 ^{x,y}Tukey test ($\alpha = 0.10$).

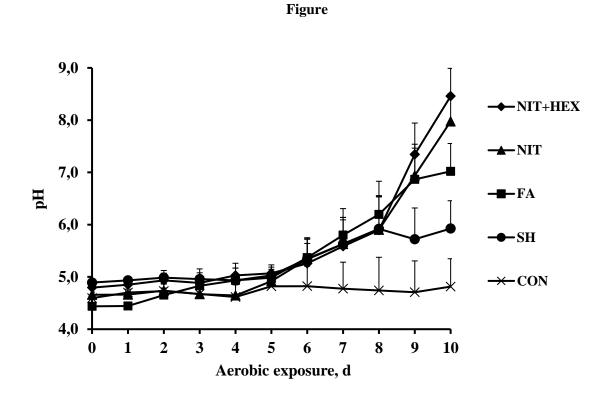




Figure 1 Development of pH in guinea grass silage during aerobic exposure for 10 d. CON: without additive, SH: soybean hulls at 100 g/kg, FA: formic acid (85%) at 4 mL/kg, NIT: sodium nitrite at 1 g/kg, NIT+HEX: sodium nitrite at 1 g/kg + hexamine at

0.65 g/kg. Bars indicate the standard error of the mean. P < 0.01 for interaction between additive treatment and day of air exposure.

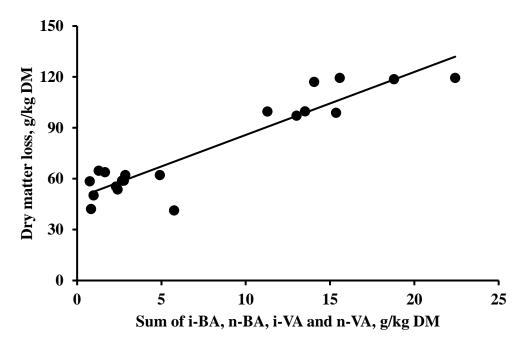


Figure 2 Relationship between the sum of i-butyric (i-BA), n-butyric (n-BA), i-valeric
(i-VA) and n-valeric (n-VA) acids (BVA, g/kg DM) and the dry mater loss (DML, g/kg

697 DM) during fermentation in guinea grass silage after 101 d of storage. DML = 48.7 +

698 3.71 × BVA, $R^2 = 0.87$, RSME = 10.4, P < 0.01.